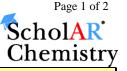
# **Calcium Hydroxide**

Section 1:



# **Product and Company Identification**

# **Calcium Hydroxide**

Synonyms/General Names: Slaked lime, Hydrated lime

Product Use: For educational use only

Manufacturer: Columbus Chemical Industries, Inc., Columbus, WI 53925.

### 24 Hour Emergency Information Telephone Numbers

CHEMTREC (USA): 800-424-9300 CANUTEC (Canada): 613-424-6666 ScholAR Chemistry; 5100 W. Henrietta Rd, Rochester, NY 14586; (866) 260-0501; www.Scholarchemistry.com

Section 2:	Hazards Identification		
White crystalline powder, no odo	r	HMIS (0 to 4	4)
		Health	1
CAUTION! Body tissue irritant	and irritating dust.	<b>Fire Hazard</b>	0
Target organs: N/A		Reactivity	0
This material is considered hazar	dous by the OSHA Hazard Communication Standard (29 CFR 1910.1200	).	

Section 3:	Composition / Information on Ingredients	
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Calcium Hydroxide (1305-62-0), <98%

Section 4:	First Aid Measures
	Always seek professional medical attention after first aid measures are provided.
Eyes:	Immediately flush eyes with excess water for 15 minutes, lifting lower and upper eyelids occasionally.
Skin:	Immediately flush skin with excess water for 15 minutes while removing contaminated clothing.
Ingestion:	Call Poison Control immediately. Rinse mouth with cold water. Give victim 1-2 cups of water or milk to drink.
-	Induce vomiting immediately.
Inhalation:	Remove to fresh air. If not breathing, give artificial respiration.
Section 5:	Fire Fighting Measures

Non-flammable solid. When heated to decomposition, emits toxic fumes.

**Protective equipment and precautions for firefighters:** Use foam or dry chemical to extinguish fire. Firefighters should wear full fire fighting turn-out gear and respiratory protection (SCBA). Cool container with water spray. Material is not sensitive to mechanical impact or static discharge.



#### Section 6:

**Accidental Release Measures** 

Use personal protection recommended in Section 8. Isolate the hazard area and deny entry to unnecessary and unprotected personnel. Sweep up spill and place in sealed bag or container for disposal. Wash spill area after pickup is complete. See Section 13 for disposal information.

# Section 7: Handling and Storage Green

Handling: Use with adequate ventilation and do not breathe dust or vapor. Avoid contact with skin, eyes, or clothing. Wash hands thoroughly after handling.

**Storage**: Store in General Storage Area [Green Storage] with other items with no specific storage hazards. Store in a cool, dry, well-ventilated, locked store room away from incompatible materials.

# Section 8:

Exposure Controls / Personal Protection

Use ventilation to keep airborne concentrations below exposure limits. Have approved eyewash facility, safety shower, and fire extinguishers readily available. Wear chemical splash goggles and chemical resistant clothing such as gloves and aprons. Wash hands thoroughly after handling material and before eating or drinking. Use NIOSH-approved respirator with a dust cartridge. Exposure guidelines: Calcium Hydroxide: OSHA PEL: 5 mg/m<sup>3</sup> (total dust)15 mg/m<sup>3</sup>(respirable fraction), ACGIH: TLV: 5 mg/m<sup>3</sup>, STEL: Not Available.

#### Section 9: **Physical and Chemical Properties**

Molecular formula	Ca(OH) <sub>2.</sub>	Appearance	White crystalline powder.
Molecular weight	74.10.	Odor	None.
Specific Gravity	2.24 g/mL @ 20°C.	Odor Threshold	N/A.
Vapor Density (air=1)	N/A.	Solubility	Sl. soluble in water, glycerin and acids.
Melting Point	580°C .	Evaporation rate	N/A (Butyl acetate = 1).
<b>Boiling Point/Range</b>	N/A.	<b>Partition Coefficient</b>	N/A $(log P_{OW})$ .
Vapor Pressure (20°C)	N/A.	рН	11, basic.
Flash Point:	N/A.	UEL	N/A.
Autoignition Temp.:	N/A.	LEL	N/A.
			N/A = Not available or applicable

# Section 10:

### **Stability and Reactivity**

Avoid heat and moisture.

Stability: Stable under normal conditions of use and storage.

Incompatibility: Acids

Shelf life: Poor shelf life, store in cool, dry environment.

# Section 11:

# **Toxicology Information**

Acute Symptoms/Signs of exposure: Eyes: Redness, tearing, itching, burning, conjunctivitis. Skin: Redness, itching. Ingestion: Irritation and burning sensations of mouth and throat, nausea, vomiting and abdominal pain. Inhalation: Irritation of mucous membranes, coughing, wheezing, shortness of breath.

Chronic Effects: No information found.

Sensitization: none expected

Calcium Hydroxide: LD50 [oral, rat]; 7340 mg/kg; LC50 [rat]; N/A; LD50 Dermal [rabbit]; N/A

Material has not been found to be a carcinogen nor produce genetic, reproductive, or developmental effects.

#### Section 12:

# **Ecological Information**

Ecotoxicity (aquatic and terrestrial): LC50 – 33.9mg/l – 96h – Clarias gariepinus.

#### Section 13:

# **Disposal Considerations**

Check with all applicable local, regional, and national laws and regulations. Local regulations may be more stringent than regional or national regulations. Small amounts of this material may be suitable for sanitary sewer or trash disposal.

Section 14:	Transport Information			
DOT Shipping Name:	Not Regulated.	Canada TDG:	Not Regulated.	
DOT Hazard Class:	N/A.	Hazard Class:	N/A.	
Identification Number:	N/A.	UN Number:	N/A.	

## Section 15:

# **Regulatory Information**

**EINECS:** Listed (215-137-3)

WHMIS Canada: E, Corrosive material, strong base.

TSCA: All components are listed or are exempt

California Proposition 65: Not listed

The product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all the information required by the Controlled Products Regulations.

#### Section 16:

#### **Other Information**

#### Current Issue Date: December 20, 2011

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